# Structure of 4-(1-Cyclohexenyl)-3-phenylsydnone* 

By J. Hašek, J. Obrda, $\dagger$ K. Huml, S. Nešpůrek and M. Šorm<br>Institute of Macromolecular Chemistry, Czechoslovak Academy of Sciences, 16206 Praha 6, Czechoslovakia

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#### Abstract

C}_{14} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2}\), monoclinic, $P 2_{1} / c, a=$ 11.015 (2), $b=10.042$ (2), $c=15.164$ (3) $\AA, \beta=$ $132.0(1)^{\circ}, V=1246 \cdot 7(4) \AA^{3}, Z=4, D_{x}=1 \cdot 29, D_{m}=$ $1 \cdot 30$ (3) $\mathrm{Mg} \mathrm{m}^{-3}, M_{r}=242 \cdot 28, \mu($ Mo $K \alpha)=0 \cdot 1$ $\mathrm{mm}^{-1}$, m.p. $397-399 \mathrm{~K}$. The structure was solved by direct methods. The final $R$ value was 0.046 for 1473 observed reflexions. The compound was studied for its photochromic properties. The angle between the planes of the phenyl and sydnone rings is $63.0^{\circ}$ and the angle between the plane of the $s p^{2}$ orbitals of the double bond $\mathrm{C}(13)=\mathrm{C}(14)$ and the sydnone ring is $28 \cdot 0^{\circ}$.


Introduction. The title compound is a member of the alkenyl sydnones, which are studied for their photochromic properties (Hašek, Obrda, Huml, Nešpůrek, Chojnacki \& Šorm, 1978; Hašek, Obrda, Huml, Nešpůrek \& Sorm, 1979). It exhibits photochromism in the crystalline state with a short half-time of decoloration and weak photochromism in solid solutions at low temperatures (Nešpůrek \& Šorm, 1977).

Bright-yellow crystals stable in air in the dark were prepared by freezing from ethanol solution. The crystal used was ground to an approximately spherical shape with a diameter varying between 0.65 and 0.75 mm . The unit-cell dimensions were obtained by refining 15 reflexions measured with a Syntex $P 2_{1}$ four-circle diffractometer. The intensities were collected by the $\theta-2 \theta$ scan with a variable scan speed of $1.5-29.5^{\circ} \mathrm{min}^{-1}$ and graphite-monochromated Mo $K \alpha$ radiation. 1639 reflexions were measured up to $\sin \theta / \lambda=0.55 \AA^{-1}$; 1473 of these were taken as observed. A reflexion was taken as observed if $I>1 \cdot 96 \sigma_{I}$, where $\sigma_{I}$ was calculated from counting statistics. The data analysis showed that the measurements were not significantly influenced by extinction or absorption. $\ddagger$

[^0]The structure was solved by direct methods with the MULTAN program (Main, Woolfson, Lessinger, Germain \& Declercq, 1974). The correct solution was indicated by the highest value of the combined figure of merit. The atomic parameters were refined by a leastsquares method in the block-diagonal approximation by the modified program NRC-10 (Ahmed, Hall, Pippy \& Huber, 1966). The function minimized was $\sum w \Delta^{2}$, where $\Delta=\left\|F_{o}|-| F_{c}\right\|, w^{-1}=\sigma_{F}^{2}+\left(0.01 F_{o}\right)^{2}\left(\sigma_{F}\right.$ determined from the counting statistics). The atomic scattering factors were taken from International Tables for $X$-ray Crystallography (1974); for $H$ the scattering factor was taken as a spherical approximation of the H atom in the hydrogen molecule (Stewart, Davidson \& Simpson, 1965).

The anisotropic refinement of non- H atoms and the isotropic refinement of the H atoms proceeded until the parameter shifts of all atoms dropped below $\frac{1}{5}$ of their e.s.d.'s. In the last refinement cycle $n=219$ atomic parameters were refined and all $m=1473$ observed reflexions were used. This led to $R_{1}=\sum \Delta / \sum\left|F_{o}\right|=$ $0.046, R_{2}=\left(\sum \Delta^{2} / \sum\left|F_{o}\right|^{2}\right)^{1 / 2}=0.045, R w_{2}=$ $\left(\sum w \Delta^{2} / \sum w\left|F_{o}\right|^{2}\right)^{1 / 2}=0.063, \quad S=\left[\sum w \Delta^{2}\right]$ $(m-n)]^{1 / 2}=4 \cdot 1$.

Table 1. Final coordinates $\left(\times 10^{4}\right)$ of the non-hydrogen atoms and their e.s.d.'s, and $B_{\mathrm{eq}}\left(\AA^{2}\right)$

| $\mathrm{O}(1)$ | $702(2)$ | $7789(2)$ | $-320(1)$ | $6 \cdot 5$ |
| :--- | ---: | ---: | ---: | ---: |
| $\mathrm{~N}(2)$ | $2145(2)$ | $8269(2)$ | $35(2)$ | $6 \cdot 1$ |
| $\mathrm{~N}(3)$ | $2747(2)$ | $8997(2)$ | $980(1)$ | $4 \cdot 5$ |
| $\mathrm{C}(4)$ | $1854(2)$ | $9027(2)$ | $1294(2)$ | $4 \cdot 2$ |
| $\mathrm{C}(5)$ | $472(3)$ | $8224(2)$ | $437(2)$ | $5 \cdot 3$ |
| $\mathrm{O}(6)$ | $-723(2)$ | $7878(2)$ | $260(1)$ | $7 \cdot 1$ |
| $\mathrm{C}(7)$ | $4317(2)$ | $9600(2)$ | $1560(2)$ | $4 \cdot 3$ |
| $\mathrm{C}(8)$ | $5661(3)$ | $9195(2)$ | $2694(2)$ | $5 \cdot 3$ |
| $\mathrm{C}(9)$ | $7171(3)$ | $9749(3)$ | $3229(2)$ | $6 \cdot 0$ |
| $\mathrm{C}(10)$ | $7301(3)$ | $668(2)$ | $2619(2)$ | $6 \cdot 0$ |
| $\mathrm{C}(11)$ | $5941(3)$ | $1043(2)$ | $1487(2)$ | $6 \cdot 1$ |
| $\mathrm{C}(12)$ | $4432(3)$ | $512(2)$ | $937(2)$ | $5 \cdot 3$ |
| $\mathrm{C}(13)$ | $2188(2)$ | $9778(2)$ | $2264(2)$ | $4 \cdot 5$ |
| $\mathrm{C}(14)$ | $1555(4)$ | $9319(3)$ | $2731(3)$ | $7 \cdot 0$ |
| $\mathrm{C}(15)$ | $1730(5)$ | $51(3)$ | $3671(3)$ | $9 \cdot 7$ |
| $\mathrm{C}(16)$ | $2912(4)$ | $1166(3)$ | $4232(2)$ | $7 \cdot 2$ |
| $\mathrm{C}(17)$ | $2885(3)$ | $1900(2)$ | $3380(2)$ | $6 \cdot 7$ |
| $\mathrm{C}(18)$ | $3120(3)$ | $1035(2)$ | $2691(2)$ | $5 \cdot 4$ |

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Table 2. Fractional coordinates of the hydrogen atoms $\left(\times 10^{3}\right)$ and $B_{\text {iso }}\left(\AA^{2}\right)$

The atoms marked by an asterisk are calculated to their ideal positions; refined positions have been deposited.

|  | $x$ | $y$ | $z$ | $B_{\text {iso }}$ |
| :---: | :---: | :---: | :---: | :---: |
| H(81) | 545 (2) | 852 (2) | 310 (2) | $6 \cdot 2$ (5) |
| H(91) | 825 (3) | 935 (2) | 406 (2) | 7.5 (6) |
| H(101) | 839 (3) | 97 (2) | 302 (2) | 7.6 (6) |
| H(111) | 604 (2) | 157 (2) | 107 (2) | $5 \cdot 8$ (5) |
| H(121) | 335 (3) | 80 (3) | 12 (2) | 8.3 (6) |
| H(141) | 79 (3) | 858 (3) | 234 (2) | 8.8 (7) |
| $\mathrm{H}(151)^{*}$ | 202 (-) | 944 (-) | 427 (-) | 15.5 (12) |
| H(152)* | 68 (-) | $37(-)$ | 335 (-) | 13.6 (10) |
| H(161)* | 399 (-) | 83 (-) | 486 (-) | 14.8 (11) |
| H(162) | 261 (3) | 177 (2) | 467 (2) | 7.5 (6) |
| H(171) | 354 (3) | 270 (3) | 371 (2) | 8.2 (6) |
| $\mathrm{H}(172)^{*}$ | 187 (-) | 238 (-) | 284 (-) | 15.5 (11) |
| H(181) | 434 (3) | 89 (3) | 327 (2) | 8.4 (6) |
| H(182) | 293 (2) | 154 (2) | 204 (2) | 5.5 (4) |



Fig. 1. Numbering scheme of 4-(1-cyclohexenyl)-3-phenylsydnone showing bond lengths $(\AA)$ and some valence angles ( ${ }^{\circ}$ ). E.s.d.'s of bond lengths are $0.003-0.006 \AA$ for non- H atoms and $0.02-$ $0.05 \AA$ for bonds involving H atoms. E.s.d.'s of $\mathrm{C}-\mathrm{C}-\mathrm{C}$ angles are $0.2-0.3^{\circ}$, of $\mathrm{C}-\mathrm{C}-\mathrm{H}$ angles $1.4-2.6^{\circ}$ and of $\mathrm{H}-\mathrm{C}-\mathrm{H}$ angles $2 \cdot 1-4 \cdot 8^{\circ}$.

The six highest maxima $\left(0.2-0.3\right.$ e $\AA^{-3}$ ) on the final difference map were located around $\mathrm{C}(14), \mathrm{C}(15)$, $\mathrm{C}(16), \mathrm{C}(17)$. Four H atoms, $\mathrm{H}(151), \mathrm{H}(152), \mathrm{H}(161)$, $H(172)$, were refined to improbable positions with high $B_{\text {iso }}$ values. Neither their recalculation to theoretical positions, nor their positioning at maxima on the difference map improved their refined parameters. Therefore, the calculated positions for these H atoms are used in this paper and the refined coordinates have been deposited.* The final fractional coordinates, $B_{\text {eq }}$ (Hamilton, 1959) and $B_{\text {iso }}$ for the H atoms are listed in Tables 1 and 2.

Discussion. The bond lengths and angles are given in Fig. 1.

By comparing this structure with that of 3-phenyl-4-(3-methyl-1-buten-2-yl)sydnone (Hašek et al., 1978), statistically significant differences were found in the angles about $\mathrm{N}(3)$ and $\mathrm{C}(4)$ (see Table 3). Characteristics of the important mean planes are shown in Table 4. According to $\chi^{2}$ of the phenyl ring, it seems that the e.s.d.'s are somewhat underestimated. Nevertheless, the sydnone ring shows significant deviations from planarity. The angle of $63.0^{\circ}$ between planes of the phenyl and sydnone rings excludes their conjugation. However, the angle of $28.0^{\circ}$ between the plane of the $s p^{2}$ orbitals of the double bond $\mathrm{C}(13)=\mathrm{C}(14)$ and the sydnone ring does not reject the possibility of a slight overlapping of $\pi$ orbitals. In connexion with this angle, short intramolecular contacts are observed between $\mathrm{C}(14) \cdots \mathrm{O}(6)[3 \cdot 138$ (3) $\AA$ ] and $\mathrm{H}(141) \cdots \mathrm{O}(6)$ [2.49 (2) $\AA$ ]; $\angle \mathrm{O}(6) \cdots \mathrm{H}(141)-\mathrm{C}(14)$ is $124(2)^{\circ}$ and $\angle \mathrm{C}(5)-\mathrm{O}(6) \cdots \mathrm{H}(141)$ is $86 \cdot 0(9)^{\circ}$.

A figure showing the molecule viewed along the $C(4)-C(13)$ bond has been deposited.* Six torsion angles formed by bonds in the cyclohexenyl ring, starting with $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16), \quad \mathrm{C}(14)-$ $C(15)-C(16)-C(17)$, etc., are $11 \cdot 7,-37 \cdot 5,54 \cdot 0$, $-43 \cdot 0,16 \cdot 4$, and $-1 \cdot 1^{\circ}$.

The thermal motion of molecule was studied by the TLS method of Schomaker \& Trueblood (1968). Part of the molecule consisting of the sydnone ring and

> * See deposition footnote.

Table 3. Comparison of the valence angles $\left(^{\circ}\right.$ ) around $\mathrm{N}(3)$ and $\mathrm{C}(4)$ in the title compound (I) and 3-phenyl-4-(3-methyl-1-buten-2-yl)sydnone (II)

|  | (I) | (II) |
| :--- | :---: | :---: |
| $\mathrm{N}(2)-\mathrm{N}(3)-\mathrm{C}(4)$ | $116 \cdot 0(2)$ | $115 \cdot 6(1)$ |
| $\mathrm{N}(2)-\mathrm{N}(3)-\mathrm{C}(7)$ | $114 \cdot 7(2)$ | $116 \cdot 5(1)$ |
| $\mathrm{C}(4)-\mathrm{N}(3)-\mathrm{C}(7)$ | $129 \cdot 3(2)$ | $127 \cdot 9(1)$ |
| $\mathrm{N}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | $104 \cdot 7(2)$ | $105 \cdot 2(1)$ |
| $\mathrm{N}(3)-\mathrm{C}(4)-\mathrm{C}(13)$ | $127 \cdot 8(2)$ | $126 \cdot 2(1)$ |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{C}(13)$ | $127 \cdot 4(2)$ | $128 \cdot 6(2)$ |

Table 4. Deviations $\left(\AA \times 10^{3}\right)$ from weighted mean planes

Asterisks denote atoms defining the planes.
Plane $a /$ phenyl ring: $\quad \chi^{2}=10$
Plane $b /$ sydnone ring: $\chi^{2}=21$
Plane $c /$ double bond: $\chi^{2}=204$

|  | Plane $a$ | Plane $b$ | Plane $c$ |
| :--- | ---: | ---: | ---: |
|  | $962(2)$ | $3(2)^{*}$ | $412(2)$ |
| $\mathrm{O}(1)$ | $1147(2)$ | $-6(2)^{*}$ | $-242(2)$ |
| $\mathrm{N}(2)$ | $44(2)$ | $4(2)^{*}$ | $-433(2)$ |
| $\mathrm{N}(3)$ | $-840(2)$ | $-2(2)^{*}$ | $12(2)^{*}$ |
| $\mathrm{C}(4)$ | $-252(3)$ | $-2(2)^{*}$ | $586(3)$ |
| $\mathrm{C}(5)$ | $-558(2)$ | $-13(2)$ | $1125(2)$ |
| $\mathrm{O}(6)$ | $-5(2)^{*}$ | $-42(2)$ | $-1150(2)$ |
| $\mathrm{C}(7)$ | $5(3)^{*}$ | $-1154(2)$ | $-2401(3)$ |
| $\mathrm{C}(8)$ | $-2(3)^{*}$ | $-1207(3)$ | $-3098(3)$ |
| $\mathrm{C}(9)$ | $-1(3)^{*}$ | $-169(3)$ | $-2542(3)$ |
| $\mathrm{C}(10)$ | $-1(3)^{*}$ | $920(3)$ | $-1297(3)$ |
| $\mathrm{C}(11)$ | $4(3)^{*}$ | $999(2)$ | $-585(3)$ |
| $\mathrm{C}(12)$ | $-2150(3)$ | $52(2)$ | $-21(2)^{*}$ |
| $\mathrm{C}(13)$ | $-2662(4)$ | $-466(3)$ | $-17(4)^{*}$ |
| $\mathrm{C}(14)$ | $-4027(5)$ | $-387(4)$ | $38(5)^{*}$ |
| $\mathrm{C}(15)$ | $-4678(4)$ | $48(3)$ | $-186(4)$ |
| $\mathrm{C}(16)$ | $-4398(3)$ | $1085(3)$ | $422(3)$ |
| $\mathrm{C}(17)$ | -430 |  |  |
| $\mathrm{C}(18)$ | $-2930(3)$ | $709(2)$ | $5(3)^{*}$ |

Angles between planes: $a \wedge b 63 \cdot 0, b \wedge c$ 28.0, $a \wedge c 65 \cdot 1^{\circ}$.
$\mathrm{O}(6), \mathrm{C}(7)$ and $\mathrm{C}(13)$ can be approximated by a rigidbody model $\left\{\right.$ r.m.s. $=\left[\sum_{i k}\left(U_{i k}^{\exp }-U_{i k}^{\text {calc }}\right)^{2} / n\right]^{1 / 2}=$ 0.0018 ; e.s.d. $=\left[\sum_{i k}\left(U_{i k}^{\text {exp }}-U_{i k}^{\text {calc }}\right)^{2} /(n-s)\right]^{1 / 2}=$ 0.0008 ; (r.m.s.) $)^{\prime}=\left[\sum_{i k} \sigma^{2}\left(U_{i k}^{\text {exp }}\right) / n\right]^{1 / 2}=0.0024$, where $n=48$ is the number of vibration tensor components and $s=20$ is the number of derived quantities $\}$. Bondlength corrections (Cruickshank, 1956) lie within two e.s.d.'s.

Fig. 2 shows a projection of the crystal structure along $\mathbf{b}$. Some short intermolecular distances are $\mathrm{O}(6)-\mathrm{H}\left(162^{\mathrm{i}}\right) \quad 2.43(3), \quad \mathrm{O}(6)-\mathrm{H}\left(91^{\mathrm{ii}}\right) \quad 2.61$ (2) $\mathrm{O}(6)-\mathrm{H}\left(182^{\text {iii }}\right) 2.66(2), \mathrm{C}(5)-\mathrm{H}\left(101^{\text {iv }}\right) 2.87$ (2), and $\mathrm{C}(5)-\mathrm{C}\left(10^{\mathrm{iv}}\right) 3.409$ (3) $\AA$, where the superscripts denote atoms in the following crystallographically equivalent positions: (i) $-x, 2-y, 0.5-z$; (ii) $x-1$, $1.5-y, 0.5-z$; (iii) $-x, 2-y,-z$; (iv) $1-x, y-0.5$, $0.5-z$. The distance $\mathrm{O}(6) \cdots \mathrm{H}\left(162^{\mathrm{i}}\right)$ is shorter than the sum of the van der Waals radii (Nyburg, 1961). According to Kvick, Koetzle \& Thomas (1974), and as in the structure of 3-phenyl-4-(3-methyl-1-buten-2-yl)sydnone, this may be classified as a very weak hydro-


Fig. 2. The crystal structure viewed down b. The molecule in the lower left-hand corner drawn in thick lines corresponds to the coordinates in Tables 1 and 2.
gen bond: $\angle \mathrm{C}\left(16^{\mathrm{i}}\right)-\mathrm{H}\left(162^{\mathrm{i}}\right) \cdots \mathrm{O}(6) \quad 153(2)^{\circ}$, $C\left(16^{i}\right) \cdots O(6) 3.443(5) \AA$.

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[^0]:    * Study of Photochromic Sydnones. III.
    $\dagger$ On leave from: Geological Institute, Czechoslovak Academy of Sciences, 16500 Praha 6, Czechoslovakia.
    $\ddagger$ Full experimental details, lists of structure factors, thermal parameters, and refined fractional coordinates of four H atoms and some projections of the structure including the thermal ellipsoids have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 34542 ( 23 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

